STAGE 3 INTERNATIONAL STUDENT PRACTICE 2018





The Joint Institute for Nuclear Research in Dubna







Joint Institute for Nuclear Research Flerov Laboratory of Nuclear Reactions University Center

PROJECT TITLE

- Ge(Li) detector for energy measurements of gamma-activity
 - Study of the operation principles of X-ray detectors
 - Moseley's Law in Action
 - Alpha Spectroscopy

SPEAKERS

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SUPERVISORS

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Introduction

3 Types of semiconductor detectors



1. X-ray Ge(Li) detector





2. Ge(Li) Gamma detector



3. Si Alpha spectroscopy detector





X-rays

X-rays are emitted when outer-shell electrons fill a vacancy in the inner shell of an atom, releasing X-rays in a pattern that is "characteristic" to each element.



X-ray transitions









History of Moseley's Law



Henry G.J. Moseley's 1887 - 1915



- In his early 20's, he measured and plotted x-ray frequencies for about 40 elements of the periodic table and was described by Rutherford as his most talented student.
- Based on his experiments, this is known as Moseley's law $E = a (Z - b)^2$
- where *a* and *b* are constants depending upon the particular spectral line, E is the energy of characteristic x-ray and Z atomic number.
- Moseley volunteered for combat duty during World War I and was killed in action at the age of 27 during the attack on the Gallipoli in the Dardenelles.





Moseley Plot of characteristic X-rays

 $\sqrt{f} = a(Z - b)$

• His data Moseley plot is still standard feature of physics textbooks (Figure 1).



Moseley plot of characteristic X-rays







 \sqrt{f} is proportional to Z a particular line (say, K_a-line).



Applications of Moseley's Law

- Any discrepancy in the order of the elements in the periodic table can be removed by Moseley's law by arranging the elements according to the atomic numbers and not according to the atomic weights.
- Moseley's law has led to the discovery of new elements like hafnium (72), technetium (43), rhenium (75) etc.
- This law has been helpful in determining the atomic number of rare earths, thereby fixing their position in the periodic table.







General view of the X-ray spectrometer









Source









Preparation for measurement

- 1. Cool the detector, fill the vessel with liquid nitrogen.
- 2. Place a calibration source in the detector.
- 3. Calibrate the energy X-ray spectrometer.
- 4. Find a Z number of an unknown source.







Moseley Law in action

Element Z (Atomic No.)	Energy (keV) K(α)	Energy (keV)K(β)
Y 39	14.1	15.8
Cd 48	22.0	24.9
Sn 50	24.2	27.2
Ce 58	33.4	37.8
Eu 63	40.1	45.4

Table 1: Energies of $K(\alpha)$ and $K(\beta)$ transitions in keV and elements listed with increasing atomic number obtained from nuclear data.



Figure 4: $K(\alpha)$ and $K(\beta)$ lines fit to Moseley's Law. We confirm the functional form of the law but different values for the constants (a & b).







Spectra from X-ray detector of unknown source



Calculation and results.

E = 74.956 keVA & B = constants obtained from K α Moseley's plot.

$$E = a (Z-b)^{2}$$

$$Z = \frac{\sqrt{E}}{a} + b$$

$$= \frac{\sqrt{74.956}}{0.0115} + 4.0137$$

$$= 83.66$$

Nuclear data: Z = 83 Element is Bi







Ge(Li) semiconductor detector

Germanium detector

is a solid state detector with P-n juction structure In which the built in area is sensitive to ionizing radiation.

> It consists of:

Semiconductor detector crystal. Pre amplifier unit.

Data spectra analyzing system.











Gamma spectra calibration



Use of Nuclear Data Search

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The Lund/LBNL Nuclear Data Search

Version 2.0, February 1999

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¹ LBNL, Berkeley, USA
² Department of Physics, Lund University, Sweden

WWW Table of Radioactive Isotopes

 Radiation search

 Nuclide search

 Atomic data (X-rays and Auger electrons, very preliminary!)

 Periodic chart interface to the nuclides

 Summary drawings for A=1-277 (PDF)

 Nuclear charts (PDF, 333 kbyte)

 Database status



The data are properly referenced as given in the database status panel. Please give your <u>feedback</u> on the usefulness of and suggestions on how to improve the Tol service.

Help and instructions are given with a "pop-up help" system:

	WWW Table of Radioactive Isotopes										
Radiation search											
Energy:	± 1 keV Search										
	Type: 🔿 Alpha 💿 Gamma										
	Parent:										
T1/2:	S 🔻 - 🛛 S 💌										
	Mass number: 152 - 152										
	Z: 63 or Element: Eu										
	N:										
Sort by:	Energy, Intensity O A, Z Reset form										
	Main page Nuclide search										

Gamma spectra of unknown source



Disintegration chain of ²²⁶Ra



20 Js	Pa221 5.9 Us 9/2-	Pa222 2.9 ms	Pa223 6.5 ms	Pa224 0.79 s	Pa225 1.7 s	Pa226 1.8 m	Pa227 38.3 m (5/2-)	Pa228 22 h 3+	Pa229 1.50 d (5/2+)	Pa230 17.4 d (2-)	Pa231 32760 y 3/2-	Pa232 1.31 d (2-)	Pa233 26.967 d 3/2-	Pa234 6.70 h 4+	Pa235 24.5 m (3/2-)	Pa236 9.1 m 1(-)	Pa237 8.7 m (1/2+)
	a	α	EC,a	α	α	EC,a	EC,α	EC,a	EC,α	EC.β;α,	a,sf	EC,β·	β·	β;sf	ß	β-	β- f
19 Js	Th220 9.7 Us 0+	Th221 1.68 ms (7/2+)	Th222 2.8 ms 0+	Th223 0.60 s (5/2)+	Th224 1.05 s 0+	Th225 8.72 m (3/2)+	Th226 30.57 m 0+	Th227 18.72 d (1/2+)	Th228 1.9116 y 0+	Th229 7340 y 5/2+	Th230 7.538E+4 y 0+	Th231 25.52 h 5/2+	Th232 1.405E10 y 0+ 0.st	Th233 22.3 m 1/2+	Th234 24.10 d 0+	Th235 7.1 m (1/2+)	Th236 37.5 m 0+
	EC,a	α	α	α	α	EC,a	α	α	α	α	a,sf	β-,α	100	β-	β-	β-	β- β-
Js Is	Ac219 11.8 Us 9/2-	Ac220 26.4 ms (3-)	Ac221 52 ms (3/2-)	Ac222 5.0 s 1-	Ac223 2.10 m (5/2-)	Ac224 2.78 h 0-	Ac225 10.0 d (3/2-)	Ac226 29,37 h (1)	Ac227 21.773 y 3/2-	Ac228 6.15 h 3+	Ac229 62.7 m (3/2+)	Ac230 122 s (1+)	Ac231 7.5 m (1/2+)	Ac232 119 s (1+)	Ac233 145 s (1/2+)	Ac234 44 s	Ac235
	EC,α	α	α	EC,α	EC,α	ΕС,β',α,	а, ¹⁴ С	EC,β;α,	β;,п	β·	ß	β·	β·	β·	ß	β·	
7 5)	Ra218 25.6 Us 0+	Ra219 10 ms (7/2)+	Ra220 18 ms 0+	Ra221 28 s 5/2+	Ra222 38.0 s 0+	Ra223 11.435 d 3/2+	Ra224 3.66 d 0+	Ra225 14.9 d 1/2+	Ra226 1600 y 0+	Ra227 42.2 m 3/2+	Ra228 5.75 y 0+	Ra229 4.0 m 5/2(+)	Ra230 93 m 0+	Ra231 103 s (7/2-,1/2+)	Ra232 250 s 0+	Ra233 30 s	Ra234 30 s 0+
	α	α	α	α	α, ¹⁴ C	a,HC	α, ¹⁴ C	β-	<u>Anc</u>	β-	β	β	β·	β-	ß	β-	β-
6 Is	Fr217 22 Us 9/2-	Fr218 1.0 ms 1- *	Fr219 20 ms 9/2-	Fr220 27.4 s 1+	Fr221 4.9 m 5/2-	Fr222 14.2 m 2-	Fr223 21.8 m 3/2(•)	Fr224 3.33 m 1	Fr225 4.0 m 3/2-	Fr226 49 s 1-	Fr227 2.47 m 1/2+	Fr228 38 s 2-	Fr229 50 s	Fr230 19.1 s	Fr231 17.5 s	Fr232	146
15	a Dn216	α Dn217	α Dn219	p.α Dn210	α D=220	p [.] Dn221	p.α D-222	P	p. Dp224	p. Do325	p Da226	p Pn227	p. D. 228	p-	5	p-	
le Je	45 Us 0+	0.54 ms 9/2+	35 ms 0+	3.96 s 5/2+	55.6 s 0+	25 m 7/2(+)	3.8235 d 0+	23.2 m 7/2	107 m 0+	4.5 m 7/2-	7.4 m 0+	22.5 s	65 s 0+		144	-	
4	a	α	α 44317	α 44310	α	p.α	44331	p;α 44222	p.	p-	p	P	p.				
s	A1215 0.10 ms 9/2-	A1210 0.30 ms 1- ΕC.β.α	A(217 32.3 ms 9/2- β.α	A1218 1.5 s B.a	Α1219 56 s β.α	3.71 A	A1221 2.3 m β·	А1222 54 s	А1225 50 s β:a		140)	142	,			
3	Po214 164.3 Us 0+	Po215 1.781 ms 9/2+ 8.a	Po216 0.145 s 0+	Po217	Po218 3.10 m 0+ 6.a		136	-)	138								
2 m	Bi213 45.59 m 9/2- 8-m	Bi214 19.9 m 1- 8-a	Bi215 7.6 m	Bi216 3.6 m (1-) 6-	1,34	-											
	Pb211 36.1 m 9/2+ 3-	Pb212 10.64 h 0+ 3-	Pb213 10.2 m (9/2+) β·	Pb214 26.8 m 0+ β	$^{226}Ra \rightarrow ^{222}Rn \rightarrow ^{218}Po \rightarrow ^{214}Pb$												

Determination of the efficiency



⁶⁰Co spectra





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Now we calculate the absolute efficiency and internal efficiency by using these equations

	S = Ndet/Nem	S = Ndet/Nfl internal	
For Co-60	22.30%	53.4%	
For Ba-133	20.8%	49.86%	-

Comparison of detectors

	Ge(Li)detector	CdZn(Te)detector
Size	Huge	Very compact
Cooling System	Need	No Need
Efficiency	More	Lower
Resolution	More	Lower







Resolution and efficciency comparison





ALPHA SPECTROMETRY WITH Si DETECTOR

0 Js	Pa221 5.9 Us 9/2-	Pa222 2.9 ms	Pa223 6.5 ms	Pa224 0.79 s	Pa225 1.7 s	Pa226 1.8 m	Pa227 38.3 m (5/2-)	Pa228 22 h 3+	Pa229 1.50 d (5/2+)	Pa230 17.4 d (2-)	Pa231 32760 y 3/2-	Pa232 1.31 d (2-)	Pa233 26.967 d 3/2-	Pa234 6.70 h 4+	Pa235 24.5 m (3/2-)	Pa236 9.1 m 1(-)	Pa237 8.7 m (1/2+)
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	EC,α	α	α	EC,α	EC,α	ΕС,β',α,	а, ¹⁴ С	EC,β;α,	β;,п	β·	ß	β·	β·	β·	ß	β·	
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	α	α	α	α	α, ¹⁴ C	a,HC	α, ¹⁴ C	β-	<u>Anc</u>	β-	β	β	β·	β-	ß	β-	β-
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Silicon detector



²²⁶Ra Source

Measurement of thickness of Al-foil (d)

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Projectile 48Ca20+ 140 MeV/u 1 pnA Fragment 42 S16+ Image: 8Be Image: 8Be Image: 1200 mg/cm2 ST Stripper Image: 11_sits SI 11_sits Image: 11_sits Image: 11_sits Image: 12_sits Image: 13_sits Image: 14_sits Image: 14_sits Image: 14_sits Image: 14_sits Image: 14_sits Image: 14_sits Image: 14_sits	PROJECTILE FRAGMENT FRAGMENT Image: Constraint of the second	alculator lement Z Q He 2 2 Stable Ion mass = 4.0015 amu Ion mass = 4.0015 amu I.92178 MeV/u Energy C I.92178 MeV/u Energy Stra Angular Stra Lateral spre T.69256 MJ/C Velocity C I.92276 cm/ns Brho (for Q= 239.554 MeV/c Beta 0.0641365 Energy Remain. E-Loss Energy Remain. E-Loss Thickness MeV/u	o Si 504 micron nain 0 MeV/u s 7.69 MeV g.(sigma) 0.0037879 MeV/u ag.(sigma) 34.282 mrad (plane) ad (sigma) 0.16256 microns =Z) 0 Tm n values for material "Si" 1.75
AI Density [g/cm3] 2.702 □ calculate reactions in this material	State Dimension Angle Solid mg/cm2 & micron Calculate Gas g/cm2 & mm 0	AI 8.53 micron 1.6427 6.5731 1.1169 1.71 C9H10 100 mm 0 0 6.5731 0.00 Range and Ran	d Energy Loss to Si
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OK Cancel	Out (Slits) Calibration Image: General setting of block Calibration	Print ? Help X Quit Charge 2t	eniginessonware Eise+4950feWare

Calculations of error

$$\left(\frac{\delta x}{x}\right) = \sqrt{\sigma^2 + \left(\frac{\Delta E}{E}\right)^2}$$

$$\sigma = \sqrt{\frac{(d_1 - \bar{d})^2 + (d_2 - \bar{d})^2 + (d_3 - \bar{d})^2 + (d_4 - \bar{d})^2}{4}}$$

$$\bar{d} = \frac{d_1 + d_2 + d_3 + d_4}{4}$$







Measurement of thickness of Al-foil by balance weighting



L= 28,9 cm H= 7,8 cm M= 0,5287 g S= L*H= 225,42 cm² D= M/S= 2,345 mg/cm² ρ= 2698,4 mg/cm³ **D/ρ= 8,69 μm**









THANK YOU FOR YOUR ATTENTION.





