



Precision investigation of modern crystalline material $\text{NbC}_{0.96}$ by neutron and X-ray diffraction

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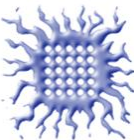
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Overview

Part 1:

Introduction to diffraction

Monochromatic beams vs. Time of Flight

Neutron vs. X-ray Diffraction

Part 2:

Neutron diffraction investigation
of $\text{NbC}_{0.96}$

Part 3:

X-ray diffraction investigation of $\text{NbC}_{0.96}$

Aim of the project

To increase our knowledge
of diffraction methods

Investigation of a real
sample

To get an experience with
Rietveld refinement

To get an experience with
Williamson-Hall method

PART-I
Introduction to Diffraction

Introduction to Diffraction

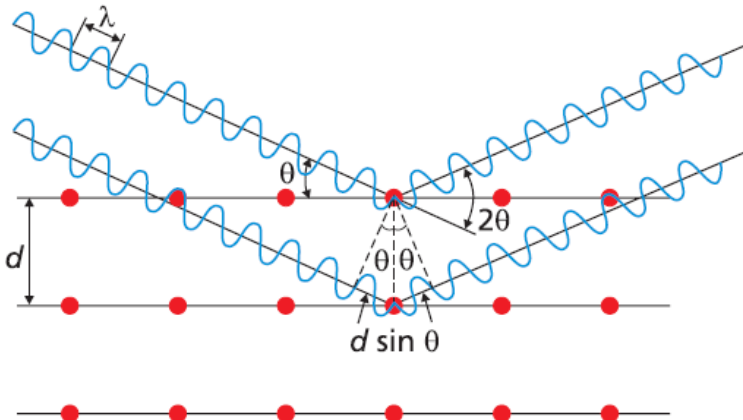
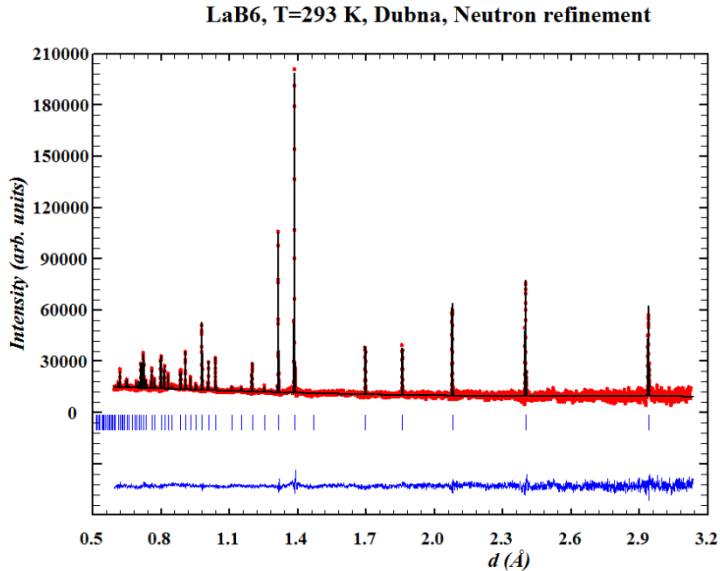
Diffraction methods are the most important approach to the analysis of crystalline solids

X-ray

electron

neutron

What can we obtain from the diffraction experiment?

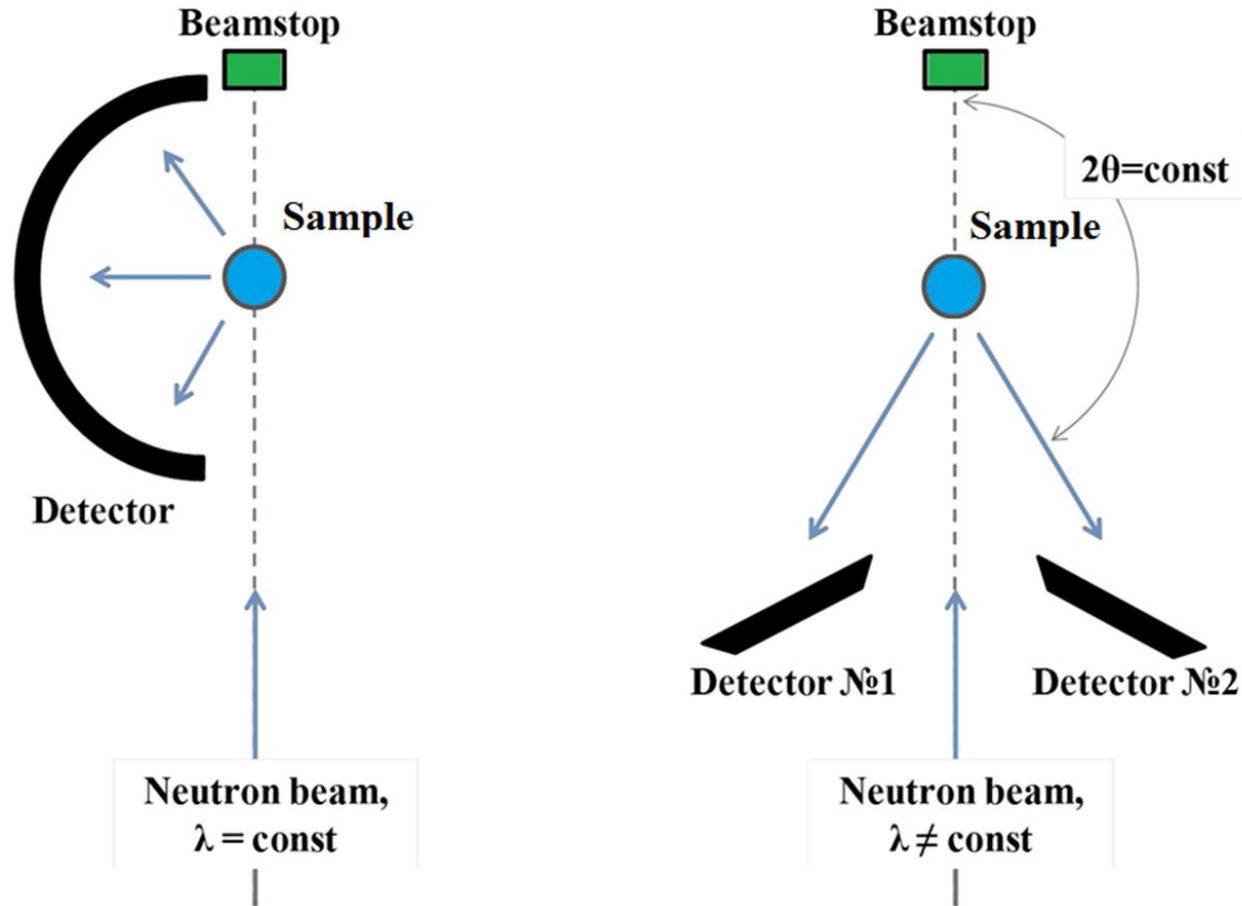


Reflection	depends on	main information
Position	<ul style="list-style-type: none"> - periodic arrangement of the atoms/molecules - wavelength used 	<ul style="list-style-type: none"> - qualitative phase analysis - lattice parameters - residual stress
Intensity	<ul style="list-style-type: none"> - crystal structure - wavelength used - sample preparation 	<ul style="list-style-type: none"> - quantitative phase analysis - crystal structure - preferred orientation/texture
Profile	<ul style="list-style-type: none"> - lattice distortions 	<ul style="list-style-type: none"> - micro-strain - crystallite sizes

Braggs law

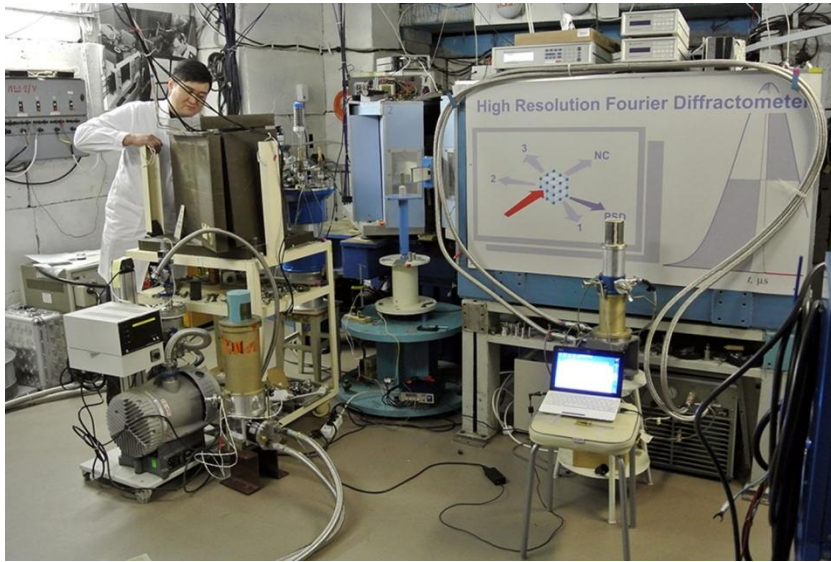
$$n\lambda = 2d \sin\theta$$

Monochromatic Beams vs. Time of Flight (TOF)



$$n\lambda = 2d \sin\theta$$

Neutron vs. X-ray Diffraction



- scattered by magnetic moments
- neutrons interact with nucleus
- scattering power independent of 2θ
- lower absorption
- large amounts of sample needed
- light elements can be seen
- low availability (nuclear reactor)



- insensitive to magnetic moments
- X-ray photons interact with electrons
- scattering power falls off with 2θ
- stronger absorption
- lower amounts of sample needed
- light elements hard to detect
- high availability (lab instrument)

PART-II
Neutron diffraction (ND)
investigation of NbC_{0.96}

Niobium Carbide



Sample used $\text{NbC}_{0.96}$

Milling times:
5, 10 and 15 h

Properties

- ceramic refractory material
- extreme hardness
- highly corrosion resistant
- high melting point

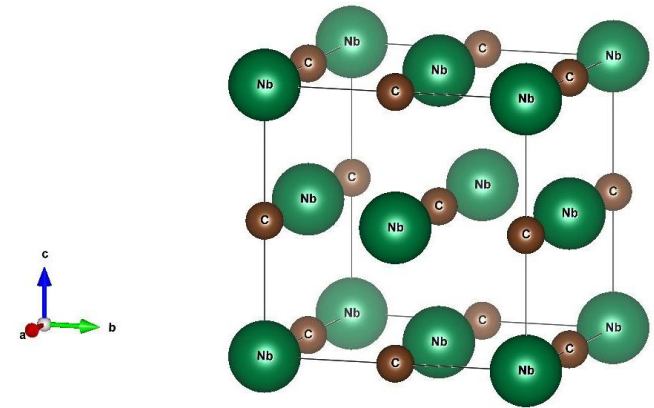
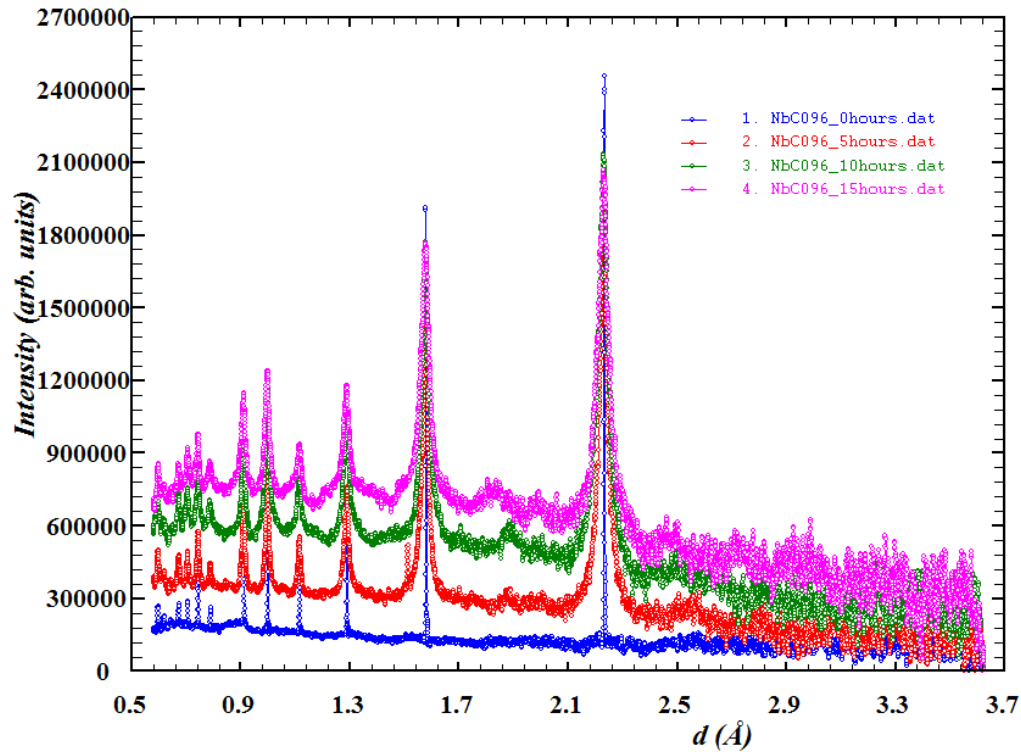
Application

- metallurgy industry
- aerospace industry
- catalysis

Neutron diffraction investigation of NbC_{0.96}

HRFD

NbC 0.96, T=293 K, Dubna, Neutron

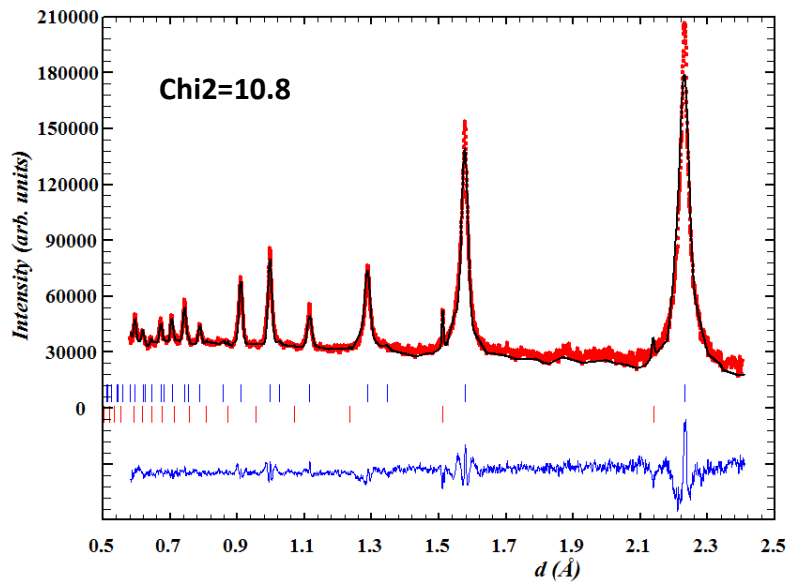


**NbC (Niobium Carbide)
powder**

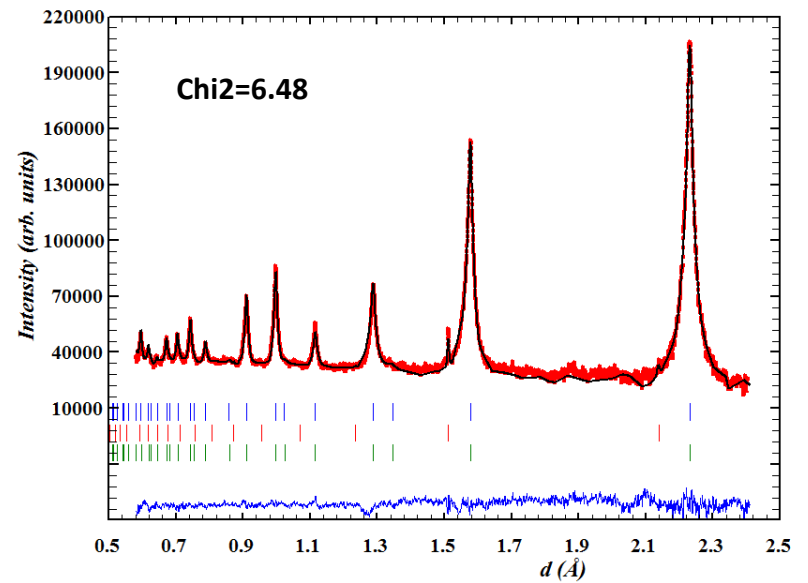
- NbC_x, x=0.96
- HRFD (high-resolution Fourier diffractometer), IBR-2
- Face-centered cubic symmetry (space group Fm3m)

Neutron spectra refinement

NbC 0.96, 5hours, T=293 K, Dubna, 1 phase

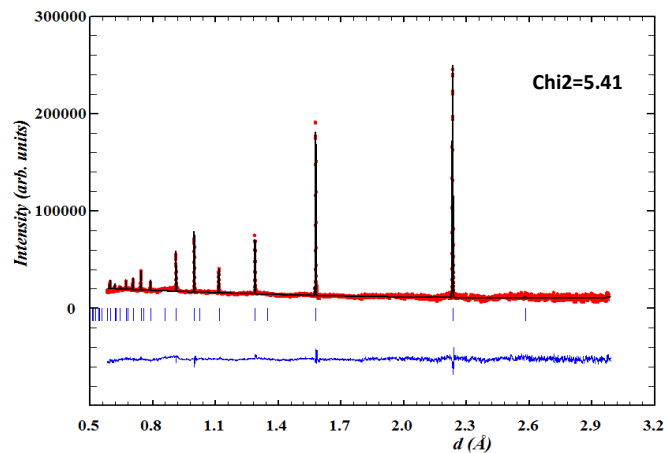


NbC 0.96, 5hours, T=293 K, Dubna, 2 phase

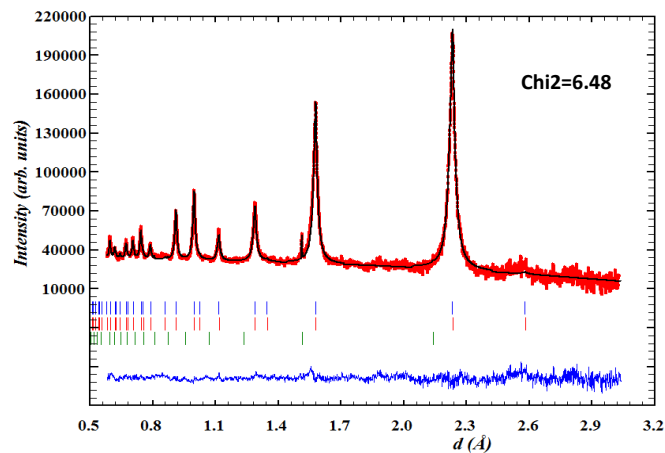


Neutron spectra refinement

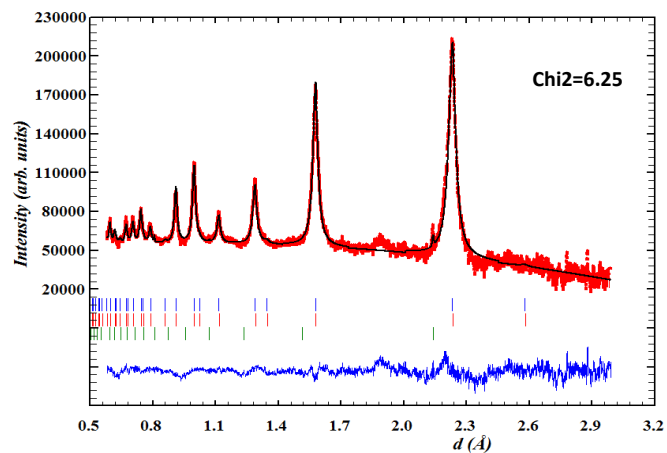
NbC 0.96 , 0 hours, T=293 K, Dubna, Neutron refinement



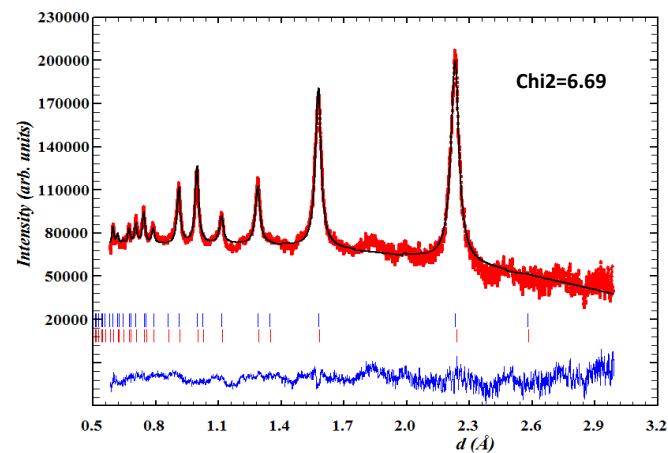
NbC 0.96, 5 hours, T=293 K, Dubna, Neutron refinement



NbC 0.96, 10 hours, T=293 K, Dubna, Neutron refinement

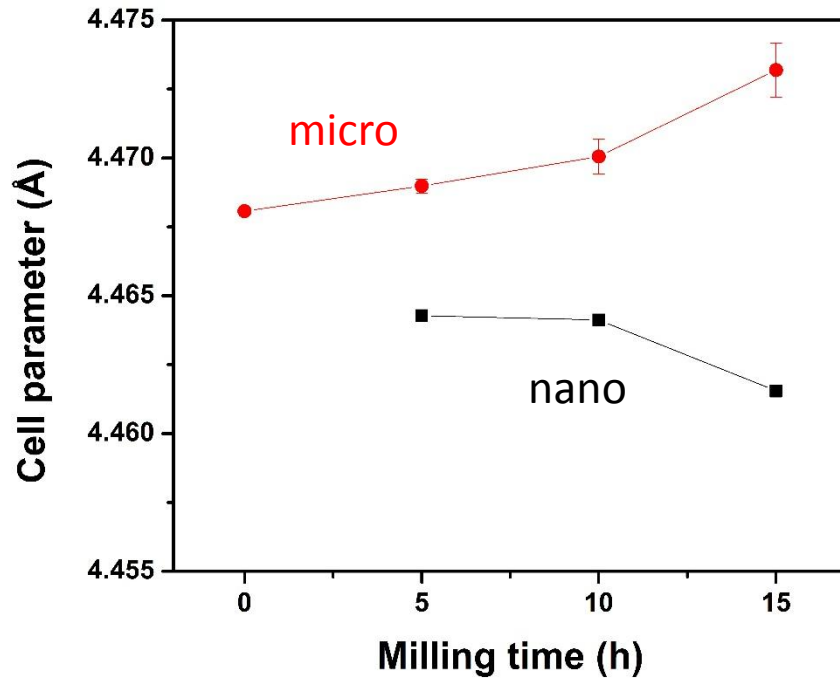


NbC 0.96, 15 hours, T=293 K, Dubna, Neutron refinement

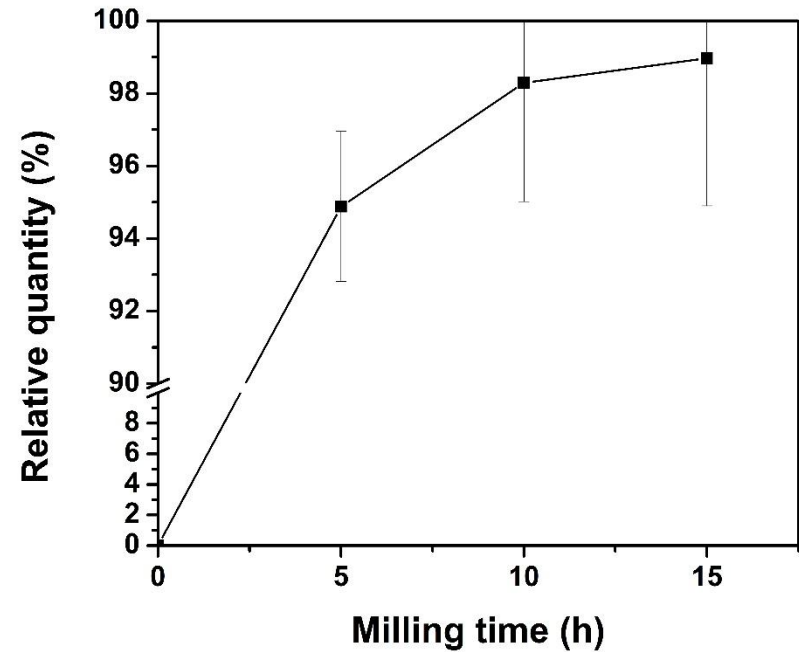


Results of Neutron spectra refinement

Dependence of the lattice parameters vs. milling time



Relative quantity of $\text{NbC}_{0.96}$ nano phase



Analysis of the widths of the NbC diffraction peaks

- Williamson-Hall method

$$W^2 = C_1 + C_2d^2 + C_3d^2 + C_4d^4 \quad (1)$$

W - FWHM obtained from the single peak fitting of the diffraction data

d - peak position

C_1, C_2 - characteristics of the diffractometer

$$C_3 = 4\varepsilon^2 \quad (2)$$

ε - microstrain

$$C_4 = \left(\frac{1}{L}\right)^2 \quad (3)$$

L - size of the grains

Average crystallite size - 20 nm

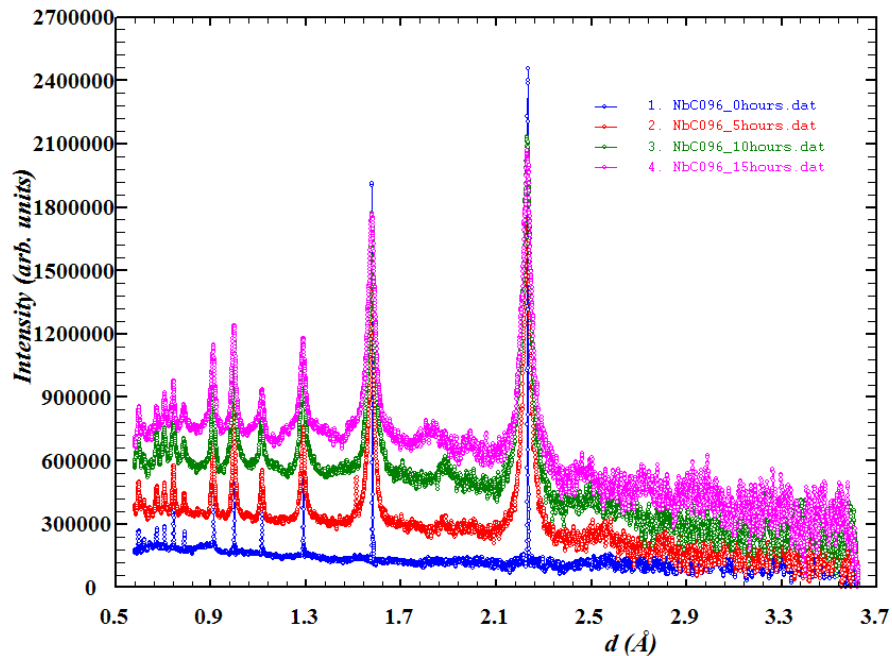
Average microstrain – 1.5 %

PART-III

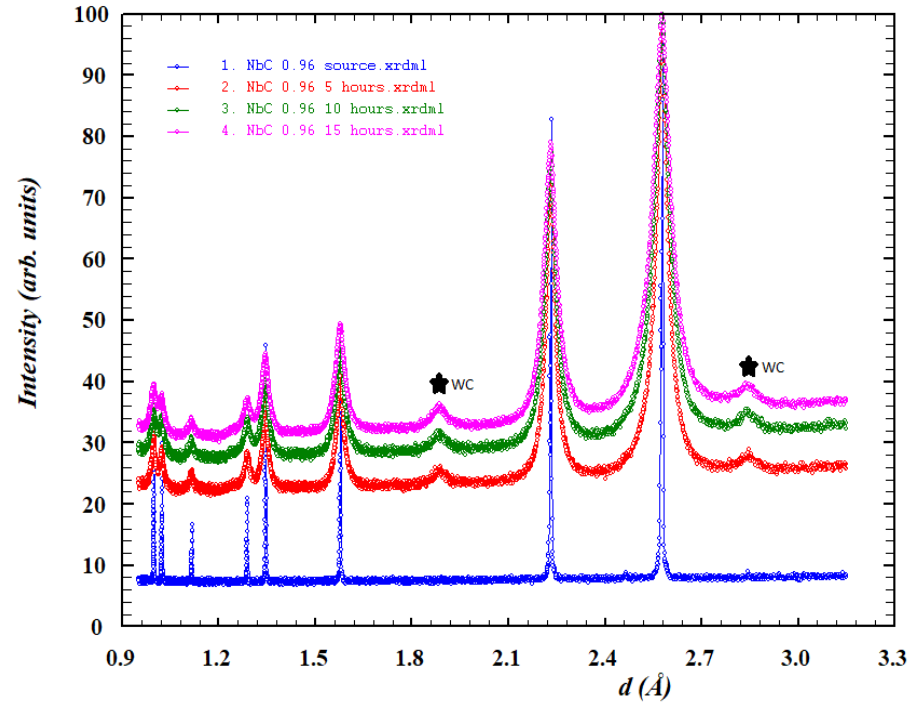
X-ray diffraction (XRD) investigation of NbC_{0.96}

XRD data vs. ND data

NbC 0.96, T=293 K, Dubna, Neutron

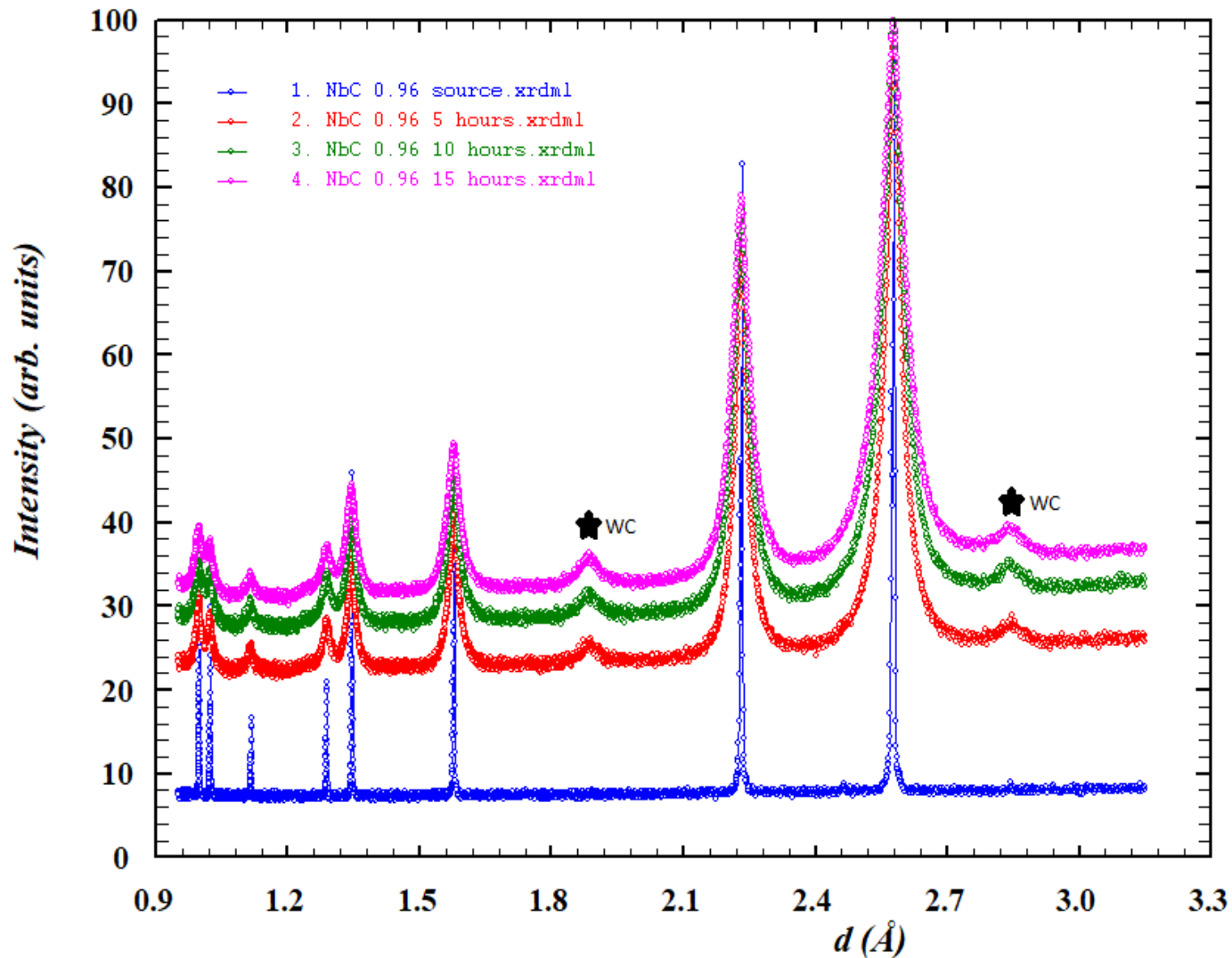


NbC 0.96, T=293K, Dubna, X-ray diff. data



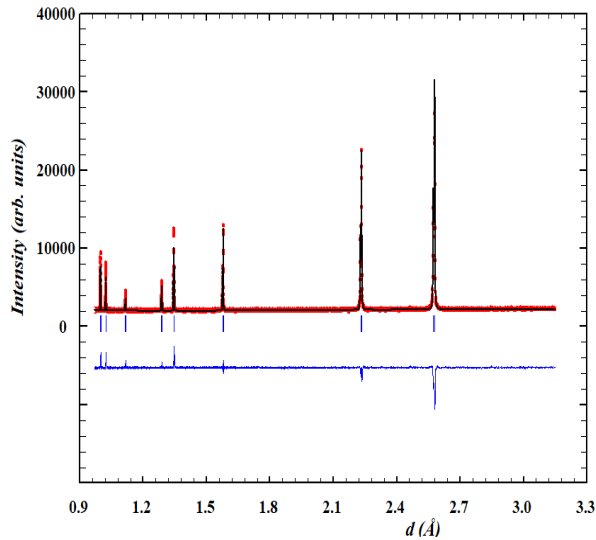
XRD investigation of NbC_{0.96}

NbC 0.96, T=293K, Dubna, X-ray diff. data

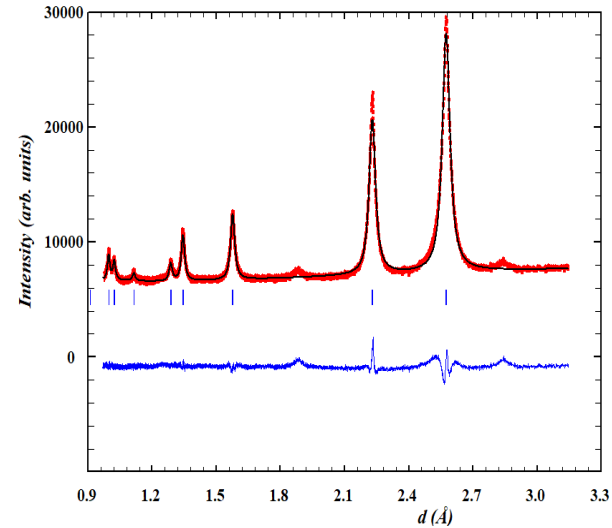


Refinement of NbC_{0.96} XRD data

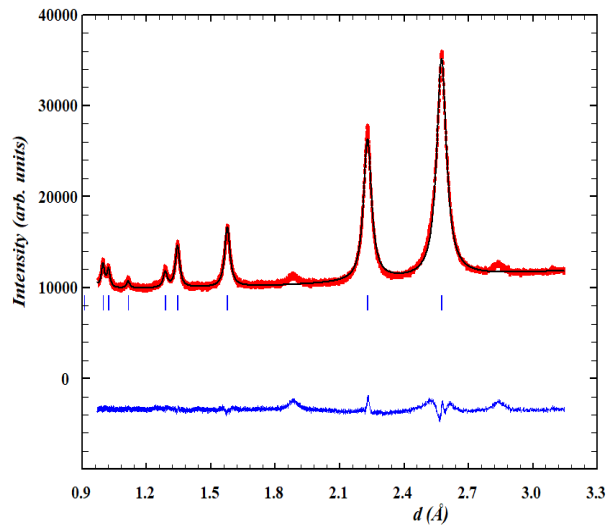
NbC 0.96, source, T=293K, Dubna, XRD refinement



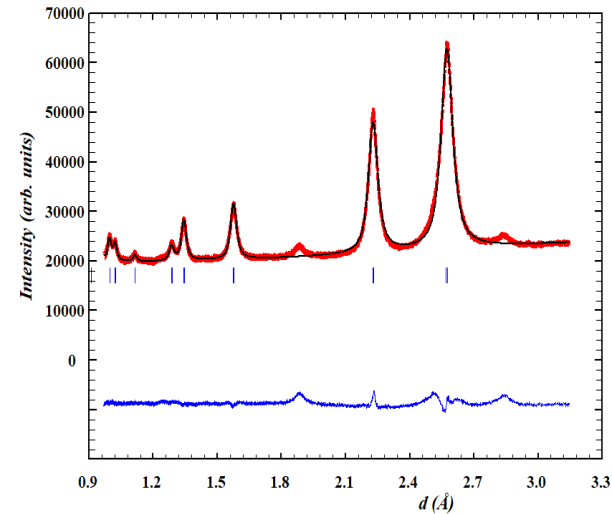
NbC 0.96, 5 hours, T=293K, Dubna, XRD refinement



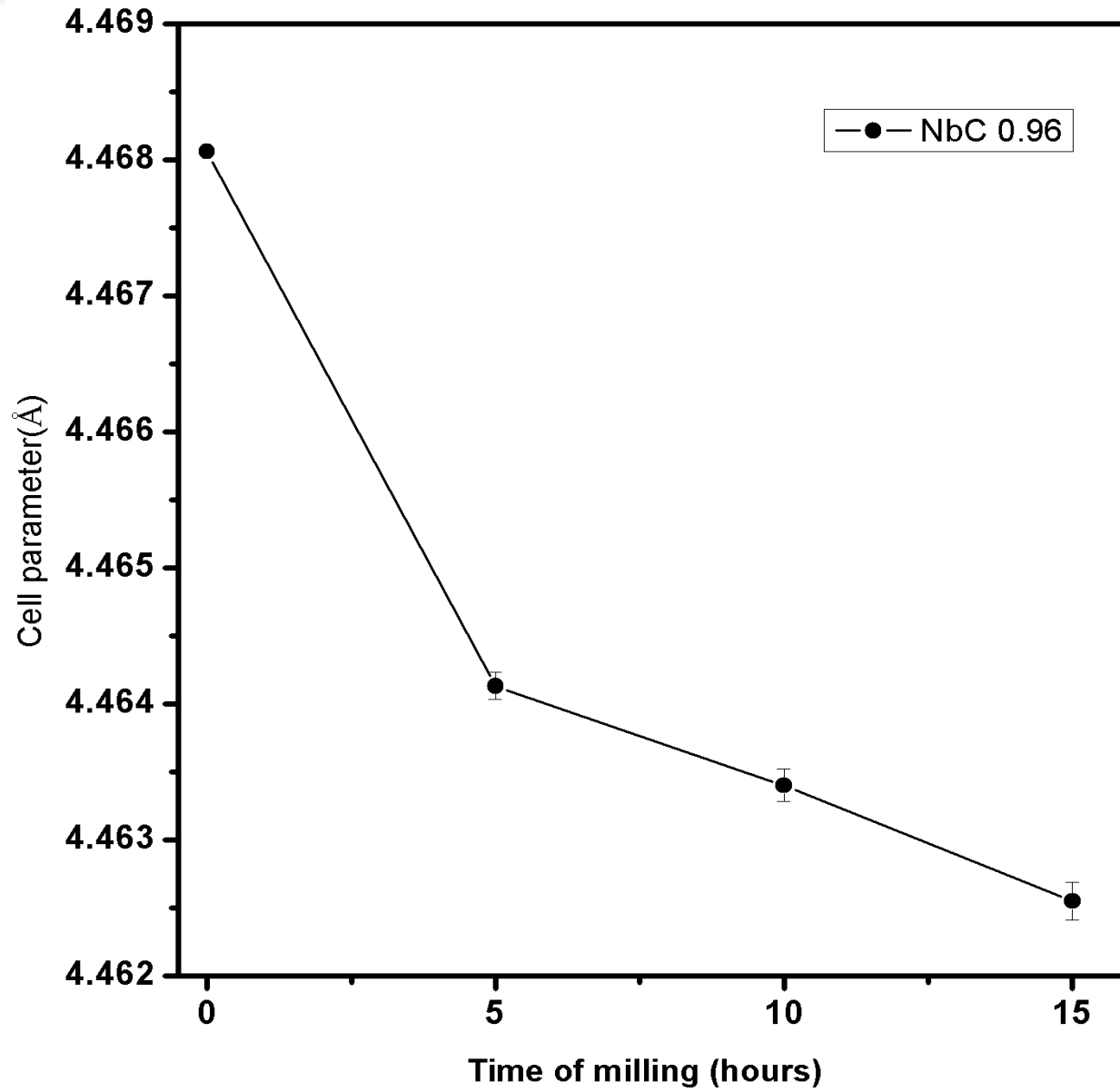
NbC 0.96, 10 hours, T=293K, Dubna, XRD refinement



NbC 0.96, 15 hours, T=293K, Dubna, XRD refinement



Cell parameter vs. Time of milling



Summary

- There is a difference in crystal structure and crystalline size between the source sample of $\text{NbC}_{0.96}$ and the milled samples.
- Using HRFD we were able to identify two fractions with different crystallite size in the sample, while we couldn't do it using our X-ray diffractometer.
- The crystallite size for nanofraction doesn't change within the error with increasing the milling time.

**Thank you for
your attention**